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St. SW., Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.

- (1)Acetone-insoluble matter (phosphatides), not less than 50.0 percent.
 - (2) Acid value, not more than 40.
- (3) Lead, not more than 1.0 part per million, as determined by atomic absorption spectroscopy.
- (4) Heavy metals (as Pb), not more than 20 parts per million.
- (5) Hexane-insoluble matter, not more than 0.3 percent.
 - (6) Peroxide value, not more than 20.
 - (7) Water, not more than 4.0 percent.
- (8) Lysolecithin, 50 to 80 mole percent of total phosphatides as determined by "Determination of Lysolecithin Content of Enzyme-Modified Lecithin: Method I," dated 1985, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the Division of Petition Control, Center for Food Safety and Applied Nutrition (HFS-215), Food and Drug Administration, 5100 Paint Branch Pkwy., College Park, MD 20740, or may be examined at the Center for Food Safety and Applied Nutrition's Library, 200 C St. SW., Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.
- (c) In accordance with §184.1(b)(1), the ingredient is used in food with no limitation other than current good manufacturing practice. The affirmation of this ingredient as generally recognized as safe as a direct human food ingredient is based upon the following current good manufacturing practice conditions of use:
- (1) The ingredient is used as an emulsifier as defined in §170.3(o)(8) of this chapter.
- (2) The ingredient is used at levels not to exceed current good manufacturing practice.

[61 FR 45889, Aug. 30, 1996]

§ 184.1065 Linoleic acid.

(a) Linoleic acid ((Z, Z)-9, 12-octadecadienoic acid ($C_{17}H_{31}COOH$) (CAS Reg. No. 60-33-3)), a straight chain unsaturated fatty acid with a molecular weight of 280.5, is a colorless oil at room temperature. Linoleic acid may be prepared from edible fats and oils by various methods including hydrolysis and saponification, the Twitchell method, low pressure splitting with catalyst, continuous high pressure counter current splitting, and medium pressure autoclave splitting with catalvst.

- (b) FDA is developing food-grade specifications for linoleic acid in cooperation with the National Academy of Sciences. In the interim, this ingredient must be of a purity suitable for its intended use. The ingredient must also meet the specifications §172.860(b) of this chapter.
- (c) In accordance with §184.1(b)(1), the ingredient is used in food with no limitation other than current good manufacturing practice. The affirmation of this ingredient as generally recognized as safe (GRAS) as a direct human food ingredient is based upon the following current good manufacturing practice conditions of use:
- (1) The ingredient is used as a flavoring agent and adjuvant as defined in §170.3(o)(12) of this chapter and as a nutrient supplement as defined in $\S 170.3(0)(20)$ of this chapter.
- (2) The ingredient is used in foods at levels not to exceed current good manufacturing practice. The ingredient may be used in infant formula in accordance with section 412(g) of the Federal Food, Drug, and Cosmetic Act (the act) or with regulations promulgated under section 412(a)(2) of the act.
- (d) Prior sanctions for this ingredient different from the uses established in this section do not exist or have been waived.

[49 FR 48534, Dec. 13, 1984]

§ 184.1069 Malic acid.

- (a) Malic acid (C₄H₆O₅ CAS Reg. No. of L-form 97-67-6, CAS Reg. No. of DLform 617-48-1) is the common name for 1-hydroxy-1, 2-ethanedicarboxylic acid. L (+) malic acid, referred to as L-malic acid, occurs naturally in various foods. Racemic DL-malic acid does not occur naturally. It is made commercially by hydration of fumaric acid or maleic
- (b) The ingredient meets the specifications of the "Food Chemicals Codex," 3d Ed. (1981), pp. 183-184, which is incorporated by reference. Copies

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may be obtained from the National Academy Press, 2101 Constitution Ave. NW., Washington, DC 20418, or may be examined at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

- (c) The ingredients are used as a flavor enhancer as defined in $\S170.3(0)(11)$ of this chapter, flavoring agent and adjuvant as defined in $\S170.3(0)(12)$ of this chapter, and pH control agent as defined in $\S170.3(0)(23)$ of this chapter.
- (d) The ingredients are used in food, except baby food, at levels not to exceed good manufacturing practice in accordance with §184.1(b)(1). Current good manufacturing practice results in a maximum level, as served, of 3.4 percent for nonalcoholic beverages as defined in §170.3(n)(3) of this chapter; 3.0 percent for chewing gum as defined in §170.3(n)(6) of this chapter; 0.8 percent for gelatins, puddings, and fillings as defined in §170.3(n)(22) of this chapter; 6.9 percent for hard candy as defined in 170.3(n)(25) of this chapter; 2.6 percent for jams and jellies as defined in §170.3(n)(28) of this chapter; 3.5 percent for processed fruits and fruit juices as defined in §170.3(n)(35) of this chapter; 3.0 percent for soft candy as defined in §170.3(n)(38) of this chapter; and 0.7 percent for all other food categories.
- (e) Prior sanctions for malic acid different from the uses established in this section do not exist or have been waived.

 $[44\ {\rm FR}\ 20656,\ {\rm Apr.}\ 6,\ 1979,\ {\rm as}\ {\rm amended}\ {\rm at}\ 49\ {\rm FR}\ 5611,\ {\rm Feb}.\ 14,\ 1984]$

§184.1077 Potassium acid tartrate.

- (a) Potassium acid tartrate ($C_4H_5KO_6$, CAS Reg. No. 868–14–4) is the potassium acid salt of L-(+)-tartaric acid and is also called potassium bitartrate or cream of tartar. It occurs as colorless or slightly opaque crystals or as a white, crystalline powder. It has a pleasant, acid taste. It is obtained as a byproduct of wine manufacture.
- (b) The ingredient meets the specifications of the Food Chemicals Codex, 3d Ed. (1981), P. 238, which is incorporated by reference. Copies are available from the National Academy Press, 2101 Constitution Ave. NW., Washington, DC 20418, or available for inspection at the Office of the Federal

Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

- (c) In accordance with §184.1(b)(1), the ingredient is used in food with no limitation other than current good manufacturing practice. The affirmation of this ingredient as generally recognized as safe (GRAS) as a direct human food ingredient is based upon the following current good manufacturing practice conditions of use:
- (1) The ingredient is used as an anticaking agent as defined in §170.3(o)(1) of this chapter; an antimicrobial agent as defined §170.3(o)(2) of this chapter; a formulation aid as defined in §170.3(o)(14) of this chapter; a humectant as defined in §170.3(o)(16) of this chapter; a leavening agent as defined in §170.3(o)(17) of this chapter; A pH control agent as defined in §170.3(o)(23) of this chapter; a processing aid as defined in §170.3(o)(24) of this chapter: a stabilizer and thickener as defined in §170.3(o)(28) of this chapter; and a surface-active agent as defined in §170.3(o)(29) of this chapter.
- (2) The ingredient is used in the following foods at levels not to exceed current good manufacturing practice: baked goods as defined in §170.3(n)(1) of this chapter; confections and frostings as defined in §170.3(n)(9) of this chapter; gelatins and puddings as defined in §170.3(n)(22) of this chapter; hard candy as defined in §170.3(n)(25) of this chapter; jams and jellies as defined in §170.3(n)(28) of this chapter; and soft candy as defined in §170.3(n)(38) of this chapter.
- (d) Prior sanctions for this ingredient different from the uses established in this section do not exist or have been waived.

[48 FR 52446, Nov. 18, 1983]

§184.1081 Propionic acid.

- (a) Propionic acid ($C_3H_6O_2$, CAS Reg. No. 79–09–4) is an oily liquid having a slightly pungent, rancid odor. It is manufactured by chemical synthesis or by bacterial fermentation.
- (b) The ingredient meets the specifications of the Food Chemicals Codex, 3d Ed. (1981), p. 254, which is incorporated by reference. Copies are available from the National Academy Press,